## I. Detergents and Cleansers

Mildly alkaline solutions of ordinary laundry soaps develop a protective coating on zinc and its alloys. They are satisfactory for both warm and cold applications. Detergent solutions are more corrosive; however, the better grades include inhibitors, which reduce corrosion to allow reasonably satisfactory service for zinc alloys. Strong alkalies (e.g., hypochlorite bleach solutions) or acid solutions (e.g., Harpic, which contains trisodium hydrogen disulfate) are to be avoided.

Table 5.1 gives results for five series of immersion trials designed to show the short-term and long-term effects of both simple chemicals and commercial solutions. Although the corrosion rates are calculated per year (for consistency with other tables), it must be remembered that often the contact time will be in hours or even minutes per year; hence, quite high corrosion rates may be acceptable. Some tests were made on zinc sheet and some on hot dip galvanized material. The results should be of the same order, so this detail is normally omitted from the tables in this chapter. When the original source shows significant differences in the two materials, it is anticipated that the dissimilarities are due to surface films (e.g., chromating) being left in place unwittingly. Trials in the United Kingdom were made by BNF and by Imperial Smelting (unpublished) at room temperature, at 70°C, and at 100°C on commercial detergents in use in Britain in 1959. Corrosion was significant at the higher temperatures, suggesting that for long life of die-cast components in washing machines, a protective coating is desirable.

## J. Trichlorethylene

Normal trichlorethylene degreasing has no effect on zinc or zinc-aluminum alloys, provided immersion in solution that has become degraded or contaminated is avoided.

## III. INORGANIC CHEMICALS

In the presence of moisture such as rain or condensation, or in solution, many chemicals react with zinc. Corrosion is serious if the reaction of the chemical with zinc forms acidic products. Mixtures of chemicals can sometimes be more corrosive than the sum of their effects taken separately but, when such mixtures come from natural sources, there are often unsuspected inhibitors present, which give a net beneficial effect.

A short immersion trial is desirable for water-based solutions. Linear extrapolation of the results will usually indicate the worst possibility, since the attack often reduces with time.

:	K L M		958) ohlavy (1957)			
	Time (days unless		Corrosion	rates		
Materials	otherwise indicated)	Temperature (°C)	mg/m <sup>2</sup> /year	µm/year	Remarks	Refs.
Detergents and Cleaners						
Calcium chloride						
20% solution	10	33	125	17	Dairy cleaning	I
20% solution	10	33	82	11	Dairy cleaning	I
20% solution + silicate	10	35	92	13	Dairy cleaning	I
20% solution + 0.05% Ca(OH) <sub>2</sub>	10	35	95	13	Dairy cleaning	1
20% solution + 0.17% Na <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>	10	35	80	11	Dairy cleaning	I
Lime mix						
200 g/L of 66.5% Ca(OH) <sub>2</sub> +	5	21	270	38	Dairy cleaning	I
32.5% MgO						
Sodium carbonate	1010					
sodium carbonate-zinc sheet	5 h	66	1,425	200	Dairy cleaning	1
sodium carbonate-HDG sheet	5 h	66	1,350	190	Dairy cleaning	I
Sodium hydroxide						
0.5% solution	5 h	66	3,250	450	Dairy cleaning	I
0.5% solution	5	21	171	24	Dairy cleaning	I
0.5% solution-HDG sheet	5 h	66	7,450	1,040	Dairy cleaning	I
0.5% solution-HDG sheet	5	21	440	61	Dairy cleaning	I

Table 5.1	Corrosion Data for Zinc and Zinc	c Coatings with Detergents,	Cleaners, Soaps,	Synthetic Detergents (Syndets), etc.

	M P	<ul> <li>Bablik and Be</li> <li>Bacon and Nu</li> </ul>	lohlavy (1957) atting (1952)			
	Time (days unless	T	Corrosion	rates		
Materials	otherwise indicated)	Temperature (°C)	mg/m <sup>2</sup> /year	µm/year	Remarks	Refs.
Sodium hypochlorite						
6 g/L, 236 ppm available Cl	5	21	190	110	Dairy cleaning	I
6 g/L, 236 ppm available Cl <sup>-</sup>	5	21	752	105	Dairy cleaning	I
Diversol						
7 g/L, 236 ppm available Cl <sup>-</sup>	5	21	175	25	Dairy cleaning	I
7 g/L, 236 ppm available Cl <sup>-</sup>	5	21	Gained weight		Dairy cleaning	I
4 g/L, 137 ppm available Cl <sup>-</sup>	5	21	215	30	Dairy cleaning	I
4 g/L, 137 ppm available Cl <sup>-</sup>	5	21	266	37	Dairy cleaning	I
Chloramine T, 0.6 g/L, 222 ppm available Cl <sup>-</sup>	5	21	617	86	Dairy cleaning	I
Chloramine T, 0.6 g/L, 222 ppm available Cl <sup>-</sup>	5	21	0	0	Dairy cleaning	I
Trisodium phosphate						
0.16% solution	5 h	66 -	1,059	150	Dairy cleaning	I
0.16% solution	5 h	66	47	7	Dairy cleaning	I
0.50% solution	5 h	66	876	120	Dairy cleaning	I
0.50% solution	5 h	66	Gained weight		Dairy cleaning	I
Carbon tetrachloride-benzol						
90%-10% mixture	40	RT	1,522	210	Dry cleaning	I
90%-10% mixture	38	140	Very high		Dry cleaning	I

Letter code for references cited in the table: International Nickel Co. (direct information)

Stupel and Koch (1959) Bukowiecki (1958)

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Perchlorethylene						
vapor	64	127	4,300	600	Dry cleaning	T
liquid	64	151	14,874	2,080	Dry cleaning	ĩ
Syndet components				-,		•
sodium tripolyphosphate 0.08%	7 h	95	6,883	965	)	м
tetrasodium pyrophosphate 0.085%	7 h	95	4,856	680		M
sodium perborate 0.064%	7 h	95	976	137		M
sodium carbonate 0.18%	7 h	95	625	87	Individual component of	M
sodium sulfate 0.12%	7 h	95	250	35	> syndet at concentration	M
sodium chloride 0.004%	7 h	95	488	68	present in detergent	м
sodium metasilicate 0.045%	7 h	95	650	91		M
lauryl alcohol sulfonate 0.18%	7 h	95	538	75		M
lauric acid ethanolamide 0.015%	7 h	90	187	26	J	М
sodium tripolyphosphate 0.15%	12 h	90	36,534	5,090	)	L
sodium tripolyphosphate 0.50%	12 h	90	49,640	6,950		L
tetrasodium pyrophosphate 0.15%	12 h	90	32,193	4,510		L
sodium metaphosphate 0.15%	12 h	90	35,551	4,980	Solution also contained	L
ordinary soap 0.3%	12 h	90	4,672	655	0.1% sodium perborate;	L
lauryl sulfate 0.05%	12 h	90	Gain	Gain	solution renewed every	L
sodium carbonate, tech calcined	12 h	90	2,409	340	hour	L
0.1%						
sodium metasilicate · 5H <sub>2</sub> O 0.03%	12 h	90	438	61		L
sodium sulfate cryst 0.05%	12 h	90	4,818	675 -	)	L
sodium trisulfate 0.15%	12 h	90	8,249	1,155	)	L
sodium carbonate 0.1%	12 h	90	1,022	143	Perborate-free solutions	L
sodium metasilicate · 5H <sub>2</sub> O 0.03%	12 h	90	Gain	Gain	reporate-free solutions	L
ordinary soap, 0.03%	12 h	90	4,745	665	j	L
Tetrasodium pyrophosphate				-		
0.1%	20 min	95	3,940	560	Effect of solution	М
0.1%	20 min	95	27,070	3,760		М
0.1%	80 min	95	10,840	1,525	<ul> <li>concentration and duration of immersion</li> </ul>	м
0.1%	7 h	95	4,850	685	or infinersion	м

	K L M		958) ohlavy (1957)			
	Time (days unless otherwise	Temperature	Corrosion	rates		
Materials	indicated)	(°C)	g/m <sup>2</sup> /5 cycles	µm/year	Remarks	Refs.
Tetrasodium pyrophosphate						
0.5 g/L in 0 g water		bp	6.0	(1,965)	)	K
0.5 g/L in 21 g water		bp	1.2	(391)		K
0.5 g/L in 21 g water + 1 g/L perborate		bp	1.5	(493)		K K
1.0 g/L in 0 g water		bp	11.2	(3,665)		K
1.0 g/L in 21 g water		bp	4.7	(1,540)	5 cycles, each 30 min at	K
1.0 g/L in 21 g water + 1 g/L perborate		bp	5.1	(1,670)	boiling + 15 min cooling: solution renewed after	К
2.0 g/L in 0 g water		bp	25.8	(8,460)	each cycle	K
2.0 g/L in 21 g water		bp	19.0	(6,450)	$g/m^2/5$ cycles $\times$ 328 = $\mu m/year$	K
2.0 g/L in 21 g water + 1 g/L perborate		bp	21.2	(6,935)	Aeration and agitation from boiling	К
3.0 g/L in 0 g water		bp	34.5	(11,300)		K
3.0 g/L in 21 g water		bp	27.2	(8,915)		K
3.0 g/L in 21 g water + 1 g/L perborate		bp	32.8	ر (10,745)	)	К

## Table 5.1 Continued

0.01%       20 min       95       2,630       380         0.05-1%       20 min       95       31,270       4,400       concentration and duration       M         0.08%       7h       95       6,880       965       concentration and duration       M         0.5 g/L in 0 g water       bp       5.4       (1,770)       K         0.5 g/L in 21 g water       bp       3.7       (1,210)       K         perborate       bp       1.4.8       (4,850)       K         1.0 g/L in 21 g water       bp       8.5       (2,800)       K         1.0 g/L in 21 g water       bp       36.9       (12,100)       K         perborate       90       36.9       (12,100)       K       Solution renewed after         2.0 g/L in 21 g water       bp       36.9       (12,100)       K       g/m <sup>2</sup> /5 cycles × 328 = $\mu$ m/year       K         2.0 g/L in 21 g water       bp       32.6       (10,700)       K       K         3.0 g/L in 21 g water       bp       32.6       (10,700)       K       K         3.0 g/L in 21 g water       bp       39.8       (13,030)       K       K         9.0 g/L in 21 g water       1 g/L       bp<
$0.08\%$ 7h       95 $6,880$ 965       of immersion       M $0.5 g/L$ in 0 g water       bp $5.4$ $(1,770)$ K $0.5 g/L$ in 21 g water       bp $2.8$ $(920)$ K $0.5 g/L$ in 21 g water       bp $3.7$ $(1,210)$ K $perborate$ bp $14.8$ $(4,850)$ K $1.0 g/L$ in 0 g water       bp $8.5$ $(2,800)$ K $1.0 g/L$ in 21 g water       bp $11.4$ $(3,730)$ K         perborate       bp $36.9$ $(12,100)$ K       Solution renewed after $2.0 g/L$ in 21 g water       bp $24.9$ $(8,150)$ $gm^2/5$ cycles $\times 328 = \mu m/year$ K $2.0 g/L$ in 21 g water       bp $22.7$ $(9,730)$ K $perborate$ $3.0 g/L$ in 21 g water       bp $32.6$ $(10,700)$ K $3.0 g/L$ in 21 g water       bp $32.6$ $(10,700)$ K       K $3.0 g/L$ in 21 g water $+ 1 g/L$ bp $39.8$ $(13,030)$ K
$0.5 \ g/L \ in 0 \ g \ water$ bp $5.4$ $(1,770)$ $0.5 \ g/L \ in 21 \ g \ water$ $bp$ $2.8$ $(920)$ $0.5 \ g/L \ in 21 \ g \ water + 1 \ g/L$ $bp$ $3.7$ $(1,210)$ perborate $K$ $K$ $1.0 \ g/L \ in 21 \ g \ water$ $bp$ $14.8$ $(4,850)$ $1.0 \ g/L \ in 21 \ g \ water$ $bp$ $8.5$ $(2,800)$ $1.0 \ g/L \ in 21 \ g \ water$ $bp$ $8.5$ $(2,800)$ $1.0 \ g/L \ in 21 \ g \ water$ $bp$ $11.4$ $(3,730)$ $perborate$ $bp$ $36.9$ $(12,100)$ $2.0 \ g/L \ in 21 \ g \ water$ $bp$ $24.9$ $(8,150)$ $2.0 \ g/L \ in 21 \ g \ water$ $bp$ $29.7$ $(9,730)$ $perborate$ $g/m^2/5 \ cycles \times 328 = \mu m/year$ $K$ $3.0 \ g/L \ in 21 \ g \ water$ $bp$ $46.2$ $(15,140)$ $3.0 \ g/L \ in 21 \ g \ water$ $bp$ $32.6$ $(10,700)$ $3.0 \ g/L \ in 21 \ g \ water$ $bp$ $39.8$ $(13,030)$
$0.5 \ g/L \ in 0 \ g \ water$ bp $5.4 \ (1,770)$ K $0.5 \ g/L \ in 21 \ g \ water$ bp $2.8 \ (920)$ $(920)$ $0.5 \ g/L \ in 21 \ g \ water + 1 \ g/L$ bp $3.7 \ (1,210)$ Kperborate $(1, 0, 0) \ g/L \ in 21 \ g \ water + 1 \ g/L$ bp $14.8 \ (4,850) \ (2,800)$ K $1.0 \ g/L \ in 21 \ g \ water + 1 \ g/L$ bp $11.4 \ (3,730) \ (2,0) \ (2,0) \ (2,0) \ (2,0) \ (2,0) \ (2,0) \ (2,0) \ (2,0) \ (2,0) \ (2,0) \ (2,0) \ (2,0) \ (2,0) \$
$0.5 \ g/L \ in 21 \ g \ water$ $bp$ $2.8 \ (920)$ $(920)$ $0.5 \ g/L \ in 21 \ g \ water + 1 \ g/L$ $bp$ $3.7 \ (1,210)$ $K$ $1.0 \ g/L \ in 0 \ g \ water$ $bp$ $14.8 \ (4,850)$ $K$ $1.0 \ g/L \ in 21 \ g \ water$ $bp$ $8.5 \ (2,800)$ $K$ $1.0 \ g/L \ in 21 \ g \ water$ $bp$ $8.5 \ (2,800)$ $K$ $1.0 \ g/L \ in 21 \ g \ water$ $bp$ $11.4 \ (3,730)$ $S \ cycles, each 30 \ min \ at$ $boiling + 15 \ min \ cooling:$ $K$ $2.0 \ g/L \ in 21 \ g \ water$ $bp$ $36.9 \ (12,100)$ $2.0 \ g/L \ in 21 \ g \ water$ $bp$ $24.9 \ (8,150)$ $2.0 \ g/L \ in 21 \ g \ water$ $bp$ $29.7 \ (9,730)$ $perborate$ $g/m^2/5 \ cycles \times 328 = \mu m/year \ K$ $3.0 \ g/L \ in 21 \ g \ water$ $bp$ $46.2 \ (15,140)$ $3.0 \ g/L \ in 21 \ g \ water$ $bp$ $32.6 \ (10,700)$ $3.0 \ g/L \ in 21 \ g \ water$ $bp$ $39.8 \ (13,030)$
$0.5 \text{ g/L in 21 g water } + 1 \text{ g/L}$ bp $3.7$ $(1,210)$ K $1.0 \text{ g/L in 0 g water}$ bp $14.8$ $(4,850)$ K $1.0 \text{ g/L in 21 g water}$ bp $8.5$ $(2,800)$ K $1.0 \text{ g/L in 21 g water}$ bp $8.5$ $(2,800)$ K $1.0 \text{ g/L in 21 g water}$ $1 \text{ g/L}$ bp $11.4$ $(3,730)$ Solution renewed after $2.0 \text{ g/L in 0 g water}$ bp $26.9$ $(12,100)$ K       solution renewed after $2.0 \text{ g/L in 21 g water}$ bp $24.9$ $(8,150)$ g/m²/5 cycles × $328 = \mu m/year$ K $2.0 \text{ g/L in 21 g water} + 1 \text{ g/L}$ bp $46.2$ $(15,140)$ K       Aeration and agitation from boiling       K $3.0 \text{ g/L in 21 g water}$ bp $32.6$ $(10,700)$ K       K $3.0 \text{ g/L in 21 g water} + 1 \text{ g/L}$ bp $39.8$ $(13,030)$ K
perborateImage: perborateImage: perborate1.0 g/L in 21 g waterbp14.8(4,850)1.0 g/L in 21 g waterbp8.5(2,800)1.0 g/L in 21 g water + 1 g/Lbp11.4(3,730)2.0 g/L in 0 g waterbp36.9(12,100)2.0 g/L in 21 g waterbp24.9(8,150)2.0 g/L in 21 g water + 1 g/Lbp29.7(9,730)3.0 g/L in 0 waterbp46.2(15,140)3.0 g/L in 21 g waterbp32.6(10,700)3.0 g/L in 21 g water + 1 g/Lbp39.8(13,030)
1.0 g/L in 0 g waterbp14.8 $(4,850)$ 1.0 g/L in 21 g waterbp8.5 $(2,800)$ 1.0 g/L in 21 g water + 1 g/Lbp11.4 $(3,730)$ perborate2.0 g/L in 0 g waterbp36.9 $(12,100)$ 2.0 g/L in 21 g waterbp24.9 $(8,150)$ 2.0 g/L in 21 g water + 1 g/Lbp29.7 $(9,730)$ 3.0 g/L in 0 waterbp46.2 $(15,140)$ 3.0 g/L in 21 g water + 1 g/Lbp32.6 $(10,700)$ 3.0 g/L in 21 g water + 1 g/Lbp39.8 $(13,030)$
1.0 g/L in 21 g water       bp       8.5       (2,800)       5 cycles, each 30 min at         1.0 g/L in 21 g water + 1 g/L       bp       11.4       (3,730)       5 cycles, each 30 min at         2.0 g/L in 0 g water       bp       36.9       (12,100)       5 cycles, each 30 min at         2.0 g/L in 21 g water       bp       36.9       (12,100)       6 cycle       K         2.0 g/L in 21 g water       bp       24.9       (8,150)       g/m <sup>2</sup> /5 cycles × 328 = $\mu$ m/year       K         2.0 g/L in 21 g water + 1 g/L       bp       29.7       (9,730)       K       Aeration and agitation from boiling       K         3.0 g/L in 0 water       bp       32.6       (10,700)       K       K         3.0 g/L in 21 g water       bp       39.8       (13,030)       K
1.0 g/L in 21 g water + 1 g/Lbp11.4 $(3,730)$ boiling + 15 min cooling:Kperborate2.0 g/L in 0 g waterbp36.9 $(12,100)$ solution renewed aftereach cycleK2.0 g/L in 21 g waterbp24.9 $(8,150)$ $g/m^2/5$ cycles × 328 = $\mu m/year$ K2.0 g/L in 21 g water + 1 g/Lbp29.7 $(9,730)$ K3.0 g/L in 0 waterbp46.2 $(15,140)$ K3.0 g/L in 21 g waterbp32.6 $(10,700)$ K3.0 g/L in 21 g water + 1 g/Lbp39.8 $(13,030)$ K
perboratesolution renewed after2.0 g/L in 0 g waterbp $36.9$ $(12,100)$ 2.0 g/L in 21 g waterbp $24.9$ $(8,150)$ 2.0 g/L in 21 g water + 1 g/Lbp $29.7$ $(9,730)$ 3.0 g/L in 0 waterbp $46.2$ $(15,140)$ 3.0 g/L in 21 g waterbp $32.6$ $(10,700)$ 3.0 g/L in 21 g water + 1 g/Lbp $39.8$ $(13,030)$
2.0 g/L in 21 g water       bp       24.9 (8,150) $g/m^2/5$ cycles × 328 = $\mu m/year$ K         2.0 g/L in 21 g water + 1 g/L       bp       29.7 (9,730)       Aeration and agitation from boiling       K         3.0 g/L in 0 water       bp       46.2 (15,140)       K       K         3.0 g/L in 21 g water       bp       32.6 (10,700)       K       K         3.0 g/L in 21 g water       bp       39.8 (13,030)       K
2.0 g/L in 21 g water       bp       24.9 (8,150) $g/m^2/5$ cycles × 328 = $\mu m/year$ K         2.0 g/L in 21 g water + 1 g/L       bp       29.7 (9,730)       Aeration and agitation from boiling       K         3.0 g/L in 0 water       bp       46.2 (15,140)       K       K         3.0 g/L in 21 g water       bp       32.6 (10,700)       K       K         3.0 g/L in 21 g water       bp       39.8 (13,030)       K
2.0 g/L in 21 g water + 1 g/L       bp       29.7       (9,730)       Aeration and agitation from boiling       K         perborate       3.0 g/L in 0 water       bp       46.2       (15,140)       K         3.0 g/L in 21 g water       bp       32.6       (10,700)       K         3.0 g/L in 21 g water + 1 g/L       bp       39.8       (13,030)       K
perborate         bp         46.2         (15,140)           3.0 g/L in 0 water         bp         32.6         (10,700)           3.0 g/L in 21 g water         bp         32.6         (10,700)           3.0 g/L in 21 g water + 1 g/L         bp         39.8         (13,030)
3.0 g/L in 21 g water     bp     32.6     (10,700)       3.0 g/L in 21 g water + 1 g/L     bp     39.8     (13,030)
3.0 g/L in 21 g water     bp     32.6     (10,700)       3.0 g/L in 21 g water + 1 g/L     bp     39.8     (13,030)
3.0 g/L in 21 g water + 1 g/L bp 39.8 (13,030) K
Sodium metaphosphate
0.5 g/L in 0 g water bp 5.9 (1,930) K
0.5 g/L in 21 g water bp 3.7 (1,210) K
0.5  g/L in 21 g water + 1 g/L bp 4.8 (1.570)
S cycles, each 30 min at boiling
1.0 g/L in 0 g water bp $17.1$ (5,610) + 15 min cooling; solution K
1.0 g/L in 21 g water bp 9.9 $(3,250)$ renewed after each cycle K
1.0  g/L in 21 g water + 1 g/L bro 15.9 (5.210)
perborate Acration and agitation from
2.0 g/L in 0 g water bp 32.9 (10,770) boiling K
2.0 g/L in 21 g water bp 28.1 (9,200) K
2.0 g/L in 21 g water + 1 g/L bp 33.1 (10,850)
perborate

	K L M		958) Iohlavy (1957)			
	Time (days unless otherwise	Temperature	Corrosion	rates		
Materials	indicated)	(°C)	g/m <sup>2</sup> /5 cycles	µm/year	Remarks	Refs.
3.0 g/L in 0 g water 3.0 g/L in 21 g water 3.0 g/L in 21 g water + 1 g/L perborate		bp bp bp	49.0 37.4 47.4	(16,050) (12,250) (15,550)	<ul> <li>5 cycles, each 30 min at boiling</li> <li>+ 15 min cooling; solution</li> <li>renewed after each cycle</li> <li>Aeration and agitation from boiling</li> </ul>	K K K
Tetrasodium pyrophosphate			mg/m <sup>2</sup> /year			
0.06% solution	6 h	80	8,570	1,200		Р
0.15% solution of mixture: 40% tetrasodium pyrophosphate + 30% trisodium phosphate · 12H <sub>2</sub> O + 30% sodium metasilicate · 5H <sub>2</sub> O	6 h	80	65,300	9,150		Ρ
0.15% solution of mixture: 31.2% sodium hexametaphosphate + 24.7% trisodium phosphate • 12H <sub>2</sub> O + 39.2% sodium metasilicate • 5H <sub>2</sub> O + 4.5% sodium carbonate	6 h	80	20,000	2,800		Р

Commercial Soaps and Syndets						
Syndet 1						
0.2% solution (Fab)	1 week	RT	235	33		I
0.2% solution (Fab)	7 h	60	18	<2		I
Syndet 2						
0.2% solution (Mytron)	1 week	RT	890	125		I
0.2% solution (Mytron)	7 h	60	57	8		I
0.2% solution + sodium silicate	1 week	RT	1,215	170		I
Syndet 3						
0.2% solution (OS)	1 week	RT	870	122		I
Syndet 4						
0.2% solution (Tide)	1 week	RT	400	56		I
0.2% solution (Tide)	7 h	60	71	10		I
Syndet 5 containing phosphates						
0.5% solution	20 min	95	51,500	7,215		M
0.5% solution	80 min	95	16,850	2,360	pH 10.20	M
0.5% solution	4 h	95	9,250	1,295		M
1.0% solution	20 min	95	89,350	12,520		M
0.5% solution		95	3,000	ן 415		M
0.5% solution		95	3,050	430		M
0.5% solution		95	4,535	655	5 consecutive 20 min cycles on same	M
0.5% solution		95	4,600	650	piece	M
0.5% solution		95	5,060	710		M
Syndet 6				,		
0.4% solution	80 min	95	13,600	1,905	pH 9.89	M
Soap 1						
0.8% solution, pH 10.65	80 min	95	4,355	610		M
0.8% solution, pH 10.65		95	1,050			M
0.8% solution, pH 10.65		95	685		5 consecutive 20 min cycles on same	M
0.8% solution, pH 10.65		95	370	}	piece	М
0.8% solution, pH 10.65		95	210		piece	M
0.8% solution, pH 10.65		95	105	J		М

	I = K = L = M = P =	<ul> <li>International N</li> <li>Stupel and Kox</li> <li>Bukowiecki (19)</li> <li>Bablik and Bel</li> </ul>	958) ohlavy (1957)		•	
	Time (days unless		Corrosion	rates		
Materials	otherwise indicated)	Temperature (°C)	mg/m <sup>2</sup> /year	µm/year	Remarks	Refs.
Soap 2			1.450	202		м
0.8% solution, pH 10.73	80 min	95	1,450	203 69		M
0.8% solution, pH 10.73	4 h	95	490 4,470	635		M
0.5% solution	20 min	95	6,310	890		M
2.0% solution	20 min	95		690		
Syndet 7 (U.S.)			g/m <sup>2</sup> /5 cycles			
i- 0 - motor		bp	49.0	(16,050)	]	K
in 0 g water in 12.5 g water		bp	24.8	(8,130)	5 cycles, each 30 min at boiling +	K
Syndet 8 (German)		-1			15 min cooling: solution renewed	
		bp	7.9	(2,590)	after each cycle.	K
in 0 g water in 12.5 g water		bp	13.8	(4,520)	Aeration and agitation from boiling	K
Syndet 9 (Swiss)		-1			,	
		bp	7.8	(2,540)	1	K
in 0 g water		bp	14.1	(4,620)	5 cycles, each 30 min at boiling +	K
in 12.5 g water		-1			15 min cooling: solution renewed	
Syndet 10 in soft water		bp	58.9	(19,250)	after each cycle.	K
+ 1% acid orthophosphate		bp	54.2	(17,730)	Aeration and agitation from boiling	K
+ 5% acid orthophosphate		bp	47.8	(15,650)	Actation and agriation from boning	К
+ 10% acid orthophosphate		bp	31.0	(10,160)	)	K
+ 10% acid orthophosphate		- P				

Syndet 11					
in soft water (silicates; no	bp	34.5	(11,330)		K
orthophosphate)	-		-		
+ 1% acid orthophosphate	bp	28.2	(9,250)		K
+ 5% acid orthophosphate	bp	17.9	(5,870)		K
+ 10% acid orthophosphate	bp	8.7	(2,845)		к
Syndet 12 in soft water + 1% acid orthophosphate + 5% acid orthophosphate + 10% acid orthophosphate Hard water soap 3 (U.S.) in 0 g water in 12.5 g water Hard water soap 4 (Swiss) in 0 g water in 12.5 g water	bp bp bp bp bp bp bp	22.4 20.4 19.6 17.3 7.9 13.2 2.2 6.6	(7,340) (6,680) (6,430) (5,660) (2,590) (4,320) (720) (2,160)	5 cycles, each 30 min at boiling + 15 min cooling: solution renewed after each cycle. g/m <sup>2</sup> /5 cycles × 328 = μm/year Aeration and agitation from boiling	К К К К К
Laboratory-Prepared Soap Mixtures and Syndets*			-		
Syndet I					
6 g/L in 0 g water	bp	40.2	(13,160)		ĸ
6 g/L in 21 g water	bp	28.5	(9,350)		K
6 g/L in 42 g water	bp	12.1	(3,960)		K
6  g/L + 0.3  g/L sodium metasilicate 0 g H <sub>2</sub> O	bp	36.4	(1,200)	5 cycles, each 30 min at boiling +	K
0.3  g/L sodium disilicate 0 g H <sub>2</sub> O	ha	32.9	(1.080)	15 min cooling: solution renewed	
0.3  g/L sodium trisilicate 0 g H <sub>2</sub> O	bp	27.6	(1,080)	after each cycle.	K
1.8 g/L sodium trisilicate 0 g H <sub>2</sub> O	bp	18.3	(9,040)	Aeration and agitation from boiling	K
0.3  g/L monoammonium phosphate	bp	22.2	(6,000)		K K
0 g H <sub>2</sub> O	bp	22.2	(7,260)		K
0.3 g/L monoammonium phosphate 21 g H <sub>2</sub> O	bp	14.6	(4,780) _	J	К

	К Ц М		958) ohlavy (1957)			
	Time (days unless	<b>m</b>	Corrosion	rates		
Materials	otherwise indicated)	Temperature (°C)	g/m <sup>2</sup> /5 cycles	µm/year	Remarks	Refs.
0.18 g/L sodium nitrate 0 g H <sub>2</sub> O 0.3 g/L magnesium lactate 0 g H <sub>2</sub> O 0.3 g/L magnesium lactate 21 g H <sub>2</sub> O 0.3 g/L calcium chloride 0 g H <sub>2</sub> O 0.3 g/L sodium thiosulfate 0.3 g/L sodium perborate 0 g H <sub>2</sub> O 0.3 g/L sodium perborate 21 g H <sub>0</sub> O 0.3 g/L sodium perborate 21 g H <sub>0</sub> O 0.3 g/L sodium perborate 21 g H <sub>0</sub> O 2 g/L perborate - 0 g H <sub>2</sub> O 2 g/L perborate - 0 g H <sub>2</sub> O 4 g/L perborate - 0 g H <sub>2</sub> O 10 g/L perborate - 0 g H <sub>2</sub> O 12 g/L perborate - 0 g H <sub>2</sub> O 12 g/L perborate - 0 g H <sub>2</sub> O 13 g/L perborate - 0 g H <sub>2</sub> O 20 g/L perborate - 0 g H <sub>2</sub> O 15 g/L perborate - 0 g H <sub>2</sub> O 20 g/L perborate - 0 g H <sub>2</sub> O	.18 g/L+	թը թը թը թը թը թը թը թը թը թր թր թր թր	24.1 19.3 21.0 16.0 16.8 21.7 56.2 34.7 51.9 55.8 64.8 75.1 79.0 81.2 82.0 81.9 80.4 63.7	(7,900) (6,325) (6,880) (5,230) (5,510) (7,110) (18,400) (11,380) (11,380) (11,380) (11,380) (21,200) (24,600) (26,000) (26,900) (26,900) (26,900) (26,900) (26,900)	5 cycles, each 30 min at boiling + 15 min cooling: solution renewed after each cycle. Aeration and agitation from boiling	кккккк ккккккккк

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Syndet II <sup>a</sup>		20	£ 000	910		I
6 g/L solution	12 h	20	5,800	810		-
6 g/L solution	12 h	40	13,200	1,850		L
6 g/L solution	12 h	65	27,800	3,890	Solution renewed every hour-HDG	L
6 g/L solution	12 h	90	57,300	8,030		L
6 g/L solution	12 h	90	59,500	8,330	+ Vigorous aeration and agitation	L
					here	
6 g/L solution	12 h	90	49,100	6,880	Solutions renewed every hour-zinc	L
without perborate, 5 g/L	12 h	90	5,250	735	Solutions renewed every hour-HDG	L
without perborate, 5 g/L	12 h	90	13,200	1,850	Vigorous aeration and agitation	L
Soap detergent I						
9 g/L	12 h	90	3,785	530	HDG-solutions renewed every hour	L
9 g/L	12 h	90	1,165	163	Zinc-solutions renewed every hour	L
8 g/L	12 h	90	1,380	193	HDG-solutions renewed every hour	L
Soap detergent II	12					×
	12 h	90	12,350	1,730	HDG-solutions renewed every hour	L
9 g/L			4,530	635	Zinc-solutions renewed every hour	I
9 g/L	12 h	90				
8 g/L	12 h	90	6,000	840	HDG-solutions renewed every hour	L

"Compositions of laboratory syndets and soaps:

Syndet I: sodium tripolyphosphate, 50%; sodium sulfate, 19%; dodecylbenzylsulfonate, 10%; sodium perborate · (4H<sub>2</sub>O), 8%; lauryl sulfate, 5%; sodium metasilicate, 5%; sodium chloride, 1%; sodium salt of EDTA, 0.5%; carboxymethyl cellulose, 0.5%; water, 1%.

Syndet II: lauryl sulfate. 10%; dodecylbenzylsulfonate, 17%; sodium tripolyphosphate, 30%; tetrasodium pyrophosphate, 5%; waterglass (powder), 6%; magnesium silicate, 2%; carboxymethylcellulose, 1%; sodium sulfate, 13%; sodium perborate, 16%.

Soap detergent I: ordinary soap, 40%; soda, 25%; trisodium phosphate, 11%; tetrasodium pyrophosphate, 5%; waterglass (powder), 6%; magnesium silicate, 2%; sodium perborate, 11%. Soap detergent II: ordinary soap, 35%; soda, 21%; trisodium phosphate, 5%; sodium tripolyphosphate, 20%; waterglass (powder), 6%; magnesium silicate, 2%; sodium perborate, 11%.